

*Supplementary Materials*

**Crystal Structure, Morphology, Optical and Super-Capacitor Properties of Sr<sub>x</sub>: α-Sb<sub>2</sub>O<sub>4</sub> Nanostructures**

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**Table 1S.** Elemental compositional analysis of pure SrCl<sub>3</sub>, pure Sb<sub>2</sub>O<sub>4</sub> and Sr:α-Sb<sub>2</sub>O<sub>4</sub> NS

Samples	Sb	Sr	Na	N	O	Co	K	Si	C	P	Purity
Pure SrCl <sub>2</sub> NS	0.05	97.8	1.02	-	0.09	-	0.37	-	-	-	99.33
Pure α-Sb <sub>2</sub> O <sub>4</sub> NS	96.8	0.07	1.36	-	1.09	-	0.11	-	-	-	99.43
Sr <sub>x</sub> :α-Sb <sub>2</sub> O <sub>4</sub> NS	88.74	10.22	0.17	-	1.44	-	0.31	-	-	-	98.96

**Table 2S.** Representing crystalline size, Morphology index, Relative Percentage Error, Texture coefficients of Sr:  $\alpha$ -Sb<sub>2</sub>O<sub>4</sub> NS

Samples	d-spacing (Å)	Crystallite size ( <i>D</i> )	Morphology Index	Relative Percentage error (RPE)	Texture coefficient
(2 wt. %) Sr: $\alpha$ -Sb <sub>2</sub> O <sub>4</sub> NS	2. 195 <sub>(222)</sub>	36.12	1.140	0.01935	0.935
(4 wt. %) Sr: $\alpha$ -Sb <sub>2</sub> O <sub>4</sub> NS	2. 199 <sub>(222)</sub>	37.02	1.150	0.01966	0.941
(8 wt. %) Sr: Sb <sub>2</sub> O <sub>4</sub> NS	2.254 <sub>(222)</sub>	38.54	1.149	0.01948	0.948
(10 wt. %) Sr: Sb <sub>2</sub> O <sub>4</sub> NS	2.267 <sub>(222)</sub>	43.547	1.120	0.01996	0.954

**Table 3S.** Optical absorptivity values of Pure SrCl<sub>2</sub>, pure  $\alpha$ -Sb<sub>2</sub>O<sub>4</sub> and Sr<sub>x</sub>:  $\alpha$ -Sb<sub>2</sub>O<sub>4</sub> (2 wt. %, 4 wt. %, 8 wt. % and 10 wt. %) NS.

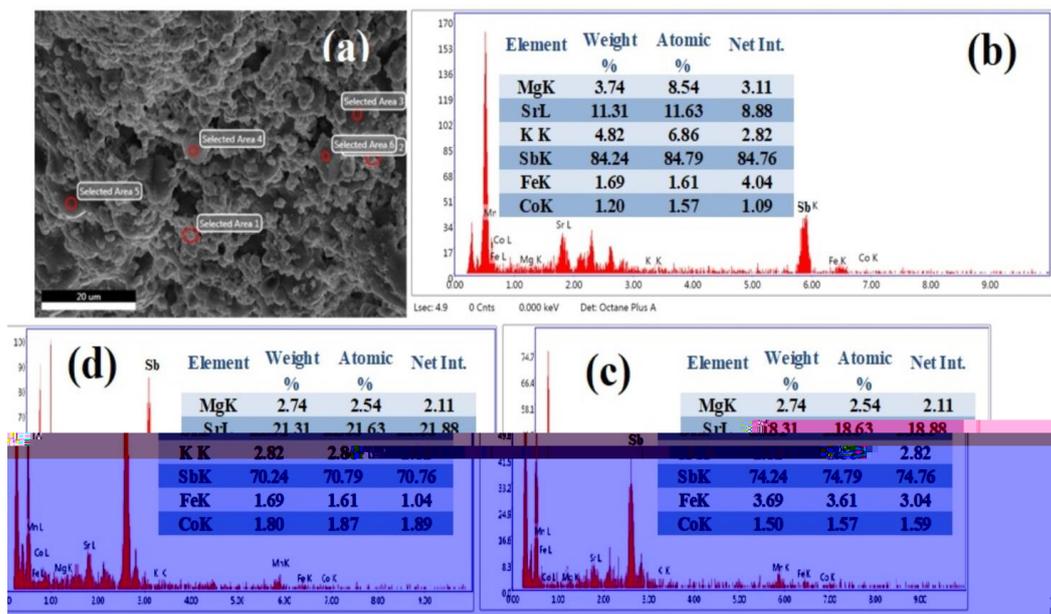
Nanostructures	$\lambda_{\text{Max}}$ (nm)	$\lambda_{\text{Shoulder 1}}$ (nm)	$\lambda_{\text{min}}$ (nm)	Absorptivity Region (nm)
Pure SrCl <sub>2</sub> NPs	380	-	-	Ultraviolet
Pure $\alpha$ -Sb <sub>2</sub> O <sub>4</sub> NPs	307	-	-	Ultraviolet
(2 wt. %) Sr: $\alpha$ -Sb <sub>2</sub> O <sub>4</sub> NS	329	350	-	Ultraviolet
(4 wt. %) Sr: $\alpha$ -Sb <sub>2</sub> O <sub>4</sub> NS	332	355	-	Ultraviolet
(8 wt. %) Sr: $\alpha$ -Sb <sub>2</sub> O <sub>4</sub> NS	343	-	391	Ultraviolet
(10wt %) Sr: $\alpha$ -Sb <sub>2</sub> O <sub>4</sub> NS	349	358	-	Ultraviolet

**Table 4S.** Band assignments, wavenumber for pure SrCl<sub>2</sub>, α-Sb<sub>2</sub>O<sub>4</sub> and Sr<sub>x</sub>: α-Sb<sub>2</sub>O<sub>4</sub> NS (2, 4, 8 and 10 wt. %)

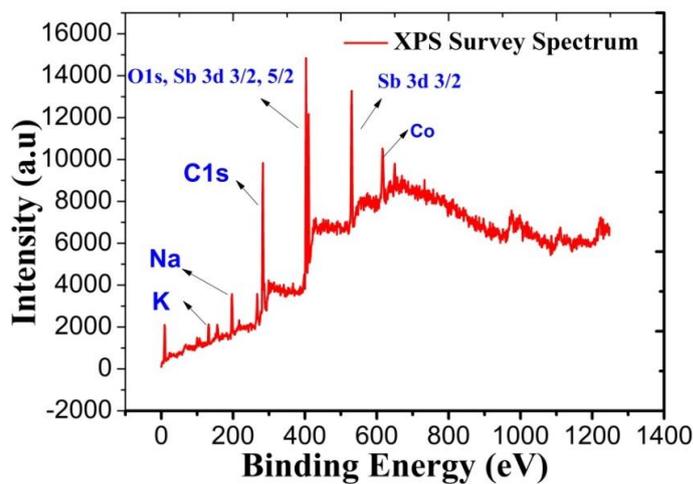
Samples	Wavenumber /cm	Band Assignments
Pure SrCl <sub>2</sub>	1522 2987	Sr-OH Stretching OH Broad Peak
Pure α-Sb <sub>2</sub> O <sub>4</sub>	690 1420 3411	Sb-O Bending C=O OH Broad Peak
2 wt. % Sr: α-Sb <sub>2</sub> O <sub>4</sub> NS	1411 1650 3121 3350	C=O Stretching Sr-O-Sr Bending Sb-OH Broad Peak OH Broad Peak
8 wt. % Sr: α-Sb <sub>2</sub> O <sub>4</sub> NS	1378 1610 2927 3395	C=O Stretching Sr-O-Sr Bending Sb-OH Broad Peak OH Broad Peak
10 wt. % Sr: α-Sb <sub>2</sub> O <sub>4</sub> NS	1371 1672 2912 3320	C=O Stretching Sr-O-Sr Bending Sb-OH Broad Peak OH Broad Peak

**Table 5S.** Specific capacitance and Rs, Rp, CPE, W of the 10 wt. % of Sr doped α-Sb<sub>2</sub>O<sub>4</sub> nanostructures

Material	Specific Capacitance (F/g) at 0.1A/g	R <sub>s</sub> (Ω)	R <sub>p</sub> (Ω)	CPE (mF)	C(mF)	W(m <sup>2</sup> Ω)
(10 wt. %) Sr: α-Sb <sub>2</sub> O <sub>4</sub> NS	890	0.978	1.95	8.02	40.3	0.514



**Figure 1S.** SEM EDX compositional analysis of (a) 10 wt. % selected area (b), (c) and (d) are the composition of various metals present in the nanostructure.



**Figure 2S.** XPS survey spectrum of Sr<sub>x</sub>: α-Sb<sub>2</sub>O<sub>4</sub> NS